The influence of molecular framework of phytochemicals from five Asteraceae species' extracts on their antibacterial and antioxidant activities

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Abstract: The structure of active compounds determines their functions. This study investigates the phytochemical composition of five plant species, tests their capacity to obstruct the growth of four pathogenic bacteria (*Bacillus subtilis, Staphylococcus aureus, Pseudomonas aeruginosa* and *Escherichia coli*), and to scavenge free radicals, and inspects the relationship between the biological activities and the phytochemical structural frameworks. GC-MS analysis was employed for specifying phytochemicals. The plant methanolic and ethanolic extracts and the antibiotic (AMC30), were evaluated for antibacterial capability utilizing the disc diffusion method. For assessing antioxidant activity, the DPPH method was used. *Achillea fragrantissima* exhibited the highest phytochemical diversity. Gram-negative bacteria were less sensitive to plant extracts. *Pulicaria undulata* and *Pulicaria incisa* demonstrated greater efficacy in suppressing bacterial proliferation. *Artemisia judaica* and *P. undulata* extracts showed higher free radicle scavenging relative to other species. The antioxidant action, antibacterial properties against *B. subtilis, S. aureus* and *E. coli* exhibited a statistically significant correlation with aromatic monocyclic compounds. A significant negative correlation was also observed between aliphatic polycyclic compound contents and *E. coli* growth inhibition, indicating that this compound category may have an *E. coli* growth stimulating capacity. This study proposes research into phytochemical frameworks and their target-specific properties.

Keywords: Asteraceae, phytochemicals, antibacterial, antioxidant, molecular frameworks.

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INTRODUCTION

Plants have a wide variety of compound classes with different molecular structures, including aromatic and aliphatic compounds, as well as monocyclic, polycyclic, and acyclic arrangements. The wide variety of phytochemicals available in nature makes them a valuable source of medicinal compounds for both humans and animals. These compounds are abundant and easily accessible. Throughout history and up to the present day, humanity has relied on phytochemicals for the prevention and treatment of numerous illnesses and health conditions. This is because many of these phytochemicals have extraordinary capabilities to fight against pathogenic bacteria, as well as other characteristics that support preventive and optimal health. The field of research investigating the relationship between chemical structure and biological activity is expanding. These studies provide valuable insights into the essential molecular components necessary for the development of potent and specific drugs. In addition, they can offer an understanding of the mechanism through which drugs produce their effects (Van Rossum 1966). Through comprehending the interplay between the structure and activity of compounds, it is possible to make targeted and precise alterations to achieve the desired function to the highest degree feasible (Lewandowski et al., 2020). The

Asteraceae family is renowned for its numerous traditional medicinal properties (Chithan et al., 2012). The genus Artemisia consists of over 400 species, the majority of which are rich in powerful compounds. The genus Pulicaria plays a significant role in traditional herbal medicine, with numerous species like P. undulata and P. incisa. The intent of the present research is to examine potential correlations between the molecular and chemical compositions of active compounds found in five medicinal plants from Asteraceae family and the biological activities of phytochemicals derived from these plants. The species were chosen based on their importance as sources of herbal medicine for the local population of Tabuk area in KSA. The results of this investigation have the potential to offer valuable understanding regarding the possible therapeutic uses of these medicinal plants and contribute to the development of new natural products for the pharmaceutical and healthcare industries. Understanding the chemical composition and biological activities of these plants may also help in preserving traditional knowledge and promoting sustainable use of herbal resources in the region.

MATERIALS AND METHODS

Plant materials

The above-ground portions of five herbaceous species from the Asteraceae family, particularly *Pulicaria*

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undulata L., Pulicaria incisa Lam., Artemisia herba-alba Asso., Artemisia judaica L., and Achillea fragrantissima Forssk., were collected from different natural vegetation sites in the Tabuk region of Northwestern Saudi Arabia. This collection was assembled during the peak growth phase of these plant species. Aside from plants gathered for phytochemical analysis, voucher specimens of these samples (No. 9 - 13) were preserved in the departmental herbarium. Then, the samples were taken to the lab to be cleansed and dried in a sheltered location. The confirmation of species identities was carried out by referring to Chaudhary (1999) and the KEW database. Subsequently, the specimens were ground and stored for further analyses.

Extraction

The extraction process followed the conventional method. Each investigated plant material, weighing 20g, was mixed with 180 ml of distilled water. The mixtures were then heated at 90°C for 30 minutes and incubated overnight at 37°C and 150 rpm in a shaking incubator (Xu *et al*, 2008). In a similar manner, round bottom flasks were employed to mix 10g of the powdered plant materials being tested with a solution of ethanol and methanol in a ratio of 9:1. The mixtures were then incubated overnight at a temperature of 37° C and a speed of 150 revolutions per minute. After the liquid extracts were separated from the solid residue through filtration using a Whatman No. 1 filter, they were concentrated using a rotary evaporator.

Phytochemical analysis

Methanol extracts were analyzed using Thermo GC Trace Ultra version 5.0 gas chromatography and an MS DSQ II mass spectrometer for GC-MS. The investigation met requirements: Helium flowed at 1 ml/min through the 30 X 0.25mm X 0.25 m DB5-MS Capillary standard nonpolar column. The oven was set at 70°C and increased by 6°C per minute to 260°C. Masses ranged from 50 to 650 m/z. It took 43 minutes to run. Classy Fire, an automated chemical categorization system, was used to classify all phytochemicals in plant extracts and determine their molecular structures and immediate ancestors (Djoumbou *et al.*, 2016). PubChem-NIH-National Library of Medicine provided the simplest synonyms for compounds (Kim *et al.*, 2023)

Antibacterial activity

The pathogenic bacterial strains selected for this study were *Bacillus subtilis* and *Staphylococcus aureus* (both Gram-positive) and *Escherichia coli* and *Pseudomonas aeruginosa* (both Gram-negative). These strains were obtained from the laboratory's stock culture and were grown on Mueller-Hinton broth agar at a temperature of 37°C for a duration of 24 hours. Prior to conducting any antimicrobial testing, the samples were subjected to subculturing following the incubation period. In order to

create inoculums, bacteria were suspended in a sterile saline solution containing 0.85% NaCl. As stated in reference [20], the suspensions were kept at an optical density (OD) of 0.4 to 0.6 at a wavelength of 405 nm. This OD value corresponds to a cell density approximately equal to 0.5 McFarland, which is equivalent to an inoculum estimated to contain 106 to 108 colony forming units per mL (CFU/mL)

Disk diffusion was employed to evaluate antibiotic susceptibility (Zazharskyi *et al.*, 2019). The method involved applying the pre-prepared samples to Mueller-Hinton agar (MHA) plates with a sterile swab. After sterilization, 6 mm Whatman paper N5 discs were immersed in 5 L of the extracts.

The soaking solution was 10% dimethyl sulfoxide and 1% tween 80 in deionized water. Under the same conditions, the extract fractions were diluted with 5g/mL Augmentin AMC30 antibiotic and 10% v/v dimethyl sulfoxide and 1% v/v tween 80 in deionized water for control. Plates were kept at room temperature before being incubated at 37° C for 24 hours. Finally, the zones of inhibition and discs were measured in millimeters to determine antibacterial efficacy.

Antioxidant activity

According to Liyana-Pathiranan et al., (2005), the 2,2diphenyl-1-picrylhydrazyl radical (DPPH) free scavenging test assessed the plants' antioxidant activity. This experiment mixed 1mL of extract with 200-1000 g/mL of 0.135mM DPPH. While gently stirring, the mixture was kept at room temperature without light for 40 Experimental positive control was ascorbic minutes. acid. The formula for DPPH scavenging activity (%) is [(Absorbance control - Absorbance sample)/Absorbance control] * 100. "Abs control" means DPPH + methanol absorbance, while "Abs sample" means DPPH radical absorbance. The absorbance of samples and control solutions was measured at 517 nm. The extracts' DPPH scavenging percentage was calculated using the equation.

Calculate DPPH scavenging activity (%) using the formula: [(Abs control - Abs sample)/Abs control] × 100. "Abs control" measures DPPH + methanol absorbance, while "Abs sample" measures DPPH radical absorbance.

The EC50, the concentration between the maximum and minimum response, was calculated using dose response models. The calculator is the 2023 Quest GraphTM EC50 Calculator from AATBioquest, Inc. The most powerful compound has the lowest EC_{50} value. In natural products chemistry, an increase in potency should likely be accompanied by an increase in a parameter that directly indicates potency. Singh *et al.* (2020) defined "effective dilution volume (EDV50)" as the inverse of the EC50 (1/EC50).

STATISTICAL ANALYSIS

PAST statistical analysis program version 4.13 was used to create a stacked chart that displays the distribution of phytochemical structural frameworks among the plant The peak areas of compounds obtained from extracts. GC-MS analysis were utilized as the foundation for determining the percentage of compounds belonging to each structural framework. This is because the peak area accurately represents the quantity of the chemical present in the extract. A bar chart was generated to display the antioxidant activity of the plants being studied. A linear Pearson correlation was carried out and a matrix was generated using JASP software version 0.18.3.0 to examine the potential correlations between the percentages of different phytochemical frameworks, antibacterial activities (measured as inhibitory zones in millimeters), and antioxidant activities (measured as EDV50 values).

RESULTS

Phytochemical composition of plant extracts

Epianastrephin (-) was the most abundant of 48 phytochemicals in *A. fragrantissima* extract (table 2 and fig. 1). *A. herba-alba* (table 2 and fig. 2) and *A. judaica* (table 3 and fig. 3) extracts contained 17 phytochemicals each, with 3-Thujanone and Xanthoxylin being the most respectively abundant. *P. incisa* extract contained 25 phytochemicals (table 4 and fig. 4), with caryophyllene oxide being the most abundant. Spathulenol was the most abundant of 24 phytochemicals in *P. undulata* extract (table 5 and fig. 5). Fig. (6) showed plant extracts with varying amounts of molecular frameworks.

The two Pulicaria species' extracts were mostly aliphatic homo-monocyclic compounds. *A. judiaca* extract was mostly aromatic homo-monocyclic compounds. However, *A. herba-alba* extract was mostly aliphatic homopolycyclic compounds. *A. fragrantissima* extract contained aliphatic and aromatic homo-monocyclic compounds. *P. incisa* and *A. herba-alba* extracts contained a few aromatic homo-monocyclic chemicals. Only the *A. judaica* extract contained aromatic heteromonocyclic compounds, while *P. undulata* contained aromatic homopolycyclic compounds. Some examples of compounds from different molecular frameworks identifies were shown in fig. 9.

Antibacterial activity

Table (6) showed that all plant extracts demonstrated higher effectiveness against gram-positive bacteria in comparison to gram-negative bacteria. The extract from P. undulata demonstrated the highest efficacy against B. subtilis, whereas the extract from P. incisa exhibited the greatest effectiveness against S. aureus. The A. judaica extract exhibited the highest efficacy against E. coli, while A. herba-alba extract was most effective against P. aeruginosa.

Antioxidant activity

EDV50 values showed that *P. undulata*, *A. judaica*, and *A. fragrantissima* extracts scavenged free radicals better than *P. incisa* and *A. herba-alba* (fig. 7).

Molecular framework/biological activity relationship

Fig. (8) demonstrated a notable positive correlation (p< 0.05) between aromatic monocyclic compounds and the inhibition ability against *E. coli*, *B. subtilis* and *S. aureus*, as well as the antioxidant activity of plant extracts. The association study uncovered a significant positive association (p<0.05) between aromatic heteromonocyclic compound content and *E. coli* bacteria inhibition. A significant negative correlation was also observed between aliphatic polycyclic compound contents and *E. coli* growth inhibition, indicating that this compound category may have an *E. coli* growth stimulating capacity.

DISCUSSION

According to previous studies, *A. fragrantissima*, among Asteraceae species, had the highest taxa_S, an alpha phytochemical diversity index that measures the number of identified compounds in the extract. Additionally, it exhibited the greatest Shannon phytochemical diversity, while *P. incisa* showed the lowest phytochemical diversity (Elbalola *et al.*, 2023). The most striking finding was that the positive control antibiotic AMC30 was inactive against *P. aeruginosa*. Gram-negative bacteria are less susceptible to antimicrobials than gram-positive bacteria because they have an external membrane.

Thus, they are expected to significantly increase global illness and death rates. Gram-negative bacteria are a major concern for the WHO (Breijyeh et al., 2020). Multiple studies have found antibiotic resistance in P. aeruginosa (Lister et al., 2009; Mulcahy et al., 2008). Ruiz-Garbajosa et al., (2017) attributed the phenomenon to the microorganism's remarkable ability to acquire common antibacterial medications. resistance to Mutations in chromosome genes or microbe-gained resistance can cause resistance. P. undulata and P. incisa have many aliphatic homo polycyclic compounds, which make them antibacterial. Spathulenol dominates P. undulata, while Bicyclo [5.2.0] nonane, 4-methylene-2,8,8trimethyl-2-vinyl- dominates P. incisa. Myriactis nepalensis essential oil contained spathulenol, a potent gram-positive bacteria inhibitor.

Many authors reported that *P. undulata* phytochemical constituents showed better free radicle scavenging compared to the other species (Rav *et al.*, 2011; Hussein 2017; Mohammed 2021). Due to their high aromatic homo monocyclic chemical content, *P. undulata*, *A. judaica* and *A. fragrantissima* are better antioxidants than other species.



Fig. 1: GC-MS chromatogram of A. fragrantissima methanolic extract.



Fig. 2: GC-MS chromatogram of *A. herba-alba* methanolic extract.



Fig. 1: GC-MS chromatogram of A. judaica methanolic extract.

No.	Compound identifier	Area	Direct parent	Molecular framework
1	cis-Carvyl acetate	739791	Menthane monoterpenoids	Aliphatic homomonocyclic
2	Verbenyl acetate (trans)	780536	bicyclic monoterpenoids	compounds Aliphatic homopolycyclic compounds
3	Thymol	4264559	Aromatic monoterpenoids,	Aromatic homomonocyclic compounds
4	(+)-trans-Isolimonene	570815	Menthane monoterpenoids	Aliphatic homomonocyclic compounds
5	Myrcenylacetat	591868	Acyclic monoterpenoids	Aliphatic acyclic compounds
6	8-Chlortheobromin	641529	Monocyclic monoterpenoids	Aliphatic heteromonocyclic compounds
7	Patchoulane	1894513	Sesquiterpenoids	Aliphatic homopolycyclic compounds
8	trans-Shisool	707843	Menthane monoterpenoids	Aliphatic homomonocyclic compounds
9	.alphaFarnesene	828854	Sesquiterpenoids	Aliphatic acyclic compounds
10	Cadabicilone	5098896	Eudesmanolides, secoeudesmanolides and derivatives	Aliphatic heteropolycyclic compounds
11	Longifolenaldehyde	1303878	Sesquiterpenoids	Aliphatic homopolycyclic
12	p-Menth-1-en-9-ol	996266	Menthane monoterpenoids	Aliphatic homomonocyclic compounds
13	Ambrosin	471385	Ambrosanolides and secoambrosanolides	Aliphatic heteropolycyclic compounds
14	IZQBEEISLWNFIO- KBPBESRZSA-N	845271	Monocyclic monoterpenoids	Aliphatic homomonocyclic compounds
15	alpha-Bulnesene	1116184	Sesquiterpenoids	Aliphatic homopolycyclic
16	2(5H)-Furanone, 5,5- dimethyl-	1084960	Butenolides	Aliphatic heteromonocyclic compounds
17	Lavender lactone	761731	Gamma butyrolactones	Aliphatic heteromonocyclic compounds
18	1,4-Cyclohexadiene, 3.3.6.6-tetramethyl-	369466	Cyclic olefins	Aliphatic homomonocyclic compounds
19	2,4-Cyclopentadiene-1- ethanamine	393449	Monoalkylamines	Aliphatic homomonocyclic compounds
20	Bicyclo[2.2.2]octenone	657382	Cyclohexenones	Aliphatic homopolycyclic compounds
21	Isophorone	848661	Cyclohexenones	Aliphatic homomonocyclic compounds
22	Ketoisophorone	546061	Cyclohexenones	Aliphatic homomonocyclic compounds
23	1-Butanone, 1-cyclohexyl-	569920	Ketones	Aliphatic homomonocyclic compounds
24	2-Cyclohexen-1-one, 3,5- dimethyl-	1150833	Cyclohexenones	Aliphatic homomonocyclic compounds
25	3,3-Dimethyl-6- methylenecyclohexene	1319543	Branched unsaturated hydrocarbons	Aliphatic homomonocyclic compounds
26	Bmlyrjoyqdjmor- uhfffaoysa-n	6277059	Branched unsaturated hydrocarbons	Aliphatic homomonocyclic compounds

Table 1: Phytochemical composition of A. fragrantissima methanolic extract, compound peak areas, direct parents and molecular frameworks.

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The influence	e of molecular	r framework of	^c phytochemical	s from five As	steraceae species'	extracts on their	· antibacterial
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27	4-ethyl-3,4- dimethylcyclohexa-2,5-dien- 1-one	395415	Cyclic ketones	Aliphatic homomonocyclic compounds
28	Lilac alcohol D	3145927	Tetrahydrofurans	Aliphatic heteromonocyclic compounds
29	2-pentynol	411645	Primary alcohols	Aliphatic acyclic compounds
30	3-Methyl-3-hexene	889484	Branched unsaturated hydrocarbons	Aliphatic acvelic compounds
31	1H-Indene, 1- ethylideneoctahydro-, trans-	441752	Branched unsaturated hydrocarbons	Aliphatic homopolycyclic compounds
32	Cyclohexanone, 2-(2-bromo- 4,4-dichlorobutyl)-	520868	Cyclic ketones	Aliphatic homomonocyclic compounds
33	3,4-Hexadienal, 2-butyl-2- ethyl-5-methyl-	2136381	Medium-chain aldehydes	Aliphatic acyclic compounds
34	Davanone B	395633	Tetrahydrofurans	Aliphatic heteromonocyclic compounds
35	Nuvzhuhadluejs-uhfffaoysa- n	433064	Branched unsaturated hydrocarbons	Aliphatic acyclic compounds
36	Ogkuljwtozzdjn-uhfffaoysa- n	402829	Primary alcohols	Aliphatic homomonocyclic compounds
37	1.5.9.13-Tetradecatetraene	454294	Alkatetraenes	Aliphatic acvelic compounds
38	Xanthoxylin	5703679	Alkyl-phenylketones	Aromatic homomonocyclic compounds
39	2,3,4-Trimethyl-hex-3-enal	722214	Medium-chain aldehydes	Aliphatic acyclic compounds
40	Lilac alcohol formate A	747131	Tetrahydrofurans	Aliphatic heteromonocyclic compounds
41	Bicyclo[6.1.0]non-1-ene	882736	Polycyclic hydrocarbons	Aliphatic homopolycyclic compounds
42	palmitic acid	522900	Long-chain fatty acids	Aliphatic acyclic compounds
43	Velleral	2351447	Organic oxides	Aliphatic homopolycyclic compounds
44	Podocarp-12-en-14-ol	834095	Hydrophenanthrenes	Aliphatic homopolycyclic compounds
45	Epianastrephin, (-)	8402652	Benzofurans	Aliphatic heteropolycyclic compounds
46	Diazoprogesterone	484466	Gluco/mineralocorticoids, progestogins and derivatives	Aliphatic homopolycyclic compounds
47	Mono(2-ethylhexyl) phthalate	4321176	Benzoic acid esters	Aromatic homomonocyclic compounds
48	Stigmasterol	434532	Stigmastanes and derivatives	Aliphatic homopolycyclic compounds

The strong positive association between antioxidant activity and the aromatic monocyclic content such as thymol, xanthoxylin, thymol acetate, alpha-Curcumene, p-Cymen-8-ol, Isoelemicin and other compounds revealed in this study supported this finding. All these phytochemicals are phenolics, which can act as antioxidants through various mechanisms (Valentão et al., 2003; Valentão et al., 2002a, 2002b, 2002c; Heim et al., 2002; Payá et al., 1992; Choi et al., 2002; Parr et al., 2002). Additionally, they can attach to metal ions that generate free radicals. Hydrophobic benzenoid rings and hydrogen bonding phenolic hydroxyl groups allow phenolics to interact with proteins. By inhibiting radicalproducing enzymes like cytochromes P450, lipoxygenases, cyclooxygenase and xanthine oxidase, they act as antioxidants.

Heterocyclic compounds are important in medical chemistry and pharmacy (Łowicki et al., 2022). Many

biologically active chemicals, including natural and pharmaceutical drugs, have cyclic organic frameworks, especially heterocycles. These frameworks are crucial in organic and medicinal chemistry. Lu *et al.* (2005) and Kusumaningsih *et al.* (2017) found that heterocyclic compounds containing nitrogen, sulfur and oxygen are potent microbe-killers.

CONCLUSIONS

Aromatic monocyclic compounds are significantly effective against E. *coli*; aliphatic hetero and homo polycyclic compound structural frameworks are the cause of the antibacterial activities against gram-positive bacteria and the aromatic homo monocyclic compound contents are the potent antioxidant agents relative to other structures.

NT.	Comment iterifier	A	D'action and	Mala sala a fue su serve ala
No.	Compound identifier	Area	Direct parent	Nolecular framework
1	3-Thujanone	21740449	Bicyclic monoterpenoids	Aliphatic homopolycyclic
2	Thuisno	17805220	Diavalia manatamanaida	A linhatia hamanalyayalia
Z	Thujone	1/803330	Bicyclic monoterpenolus	compounds
3	cis-beta-Farnesene	3632623	Sesquiterpenoids	Aliphatic acyclic compounds
4	Sesquiphellandrene	5566314	Sesquiterpenoids	Aliphatic homomonocyclic
				compounds
5	(-)-Spathulenol	1705634	5,10-cycloaromadendrane	Aliphatic homopolycyclic
			sesquiterpenoids	compounds
6	Nfzwwjwnbxicbw-uhfffaoysa-n	5988751	Eudesmane, isoeudesmane	Aliphatic homopolycyclic
			or cycloeudesmane	compounds
			sesquiterpenoids	
7	(4E)-2,7-Dimethyl-4,6-octadien-2-ol #	16584349	Tertiary alcohols	Aliphatic acyclic compounds
8	Artemisia ketone	6317033	Enones	Aliphatic acyclic compounds
9	palmitic acid	5511831	Long-chain fatty acids	Aliphatic acyclic compounds
10	(Z)-6-Dodecen-4-olide	7330029	Gamma butyrolactones	Aliphatic heteromonocyclic compounds
11	4-Methyl-1,3-pentadiene	13033701	Branched unsaturated	Aliphatic acyclic compounds
12	Icvzmtigtxbihi-clfysbassa-n	2651220	Branched unsaturated	Aliphatic acyclic compounds
12	T 1 4 11	1200205	nydrocarbons	A (* 1 1*
13	Isobutylbenzene	1280395	Phenylpropanes	Aromatic homomonocyclic compounds
14	Sbtowvndplpksy-uhfffaoysa-n	5244980	Trialkylheterosilanes	Aliphatic heteromonocyclic compounds
15	Nigpdtpduaqpha-uhfffaoysa-n	2451344	Organobromides	Aliphatic homomonocyclic
				compounds
16	10-Methylanthracene-9-carboxaldehyde	9412869	Anthracenes	Aromatic homopolycyclic compounds
17	Spathulenol	3786275	5,10-cycloaromadendrane	Aliphatic homopolycyclic
	*		sesquiterpenoids	compounds

Table 2: Phytochemical composition of A. herba-alba methanolic extract, compound peak areas, direct parents and molecular frameworks



Fig. 2: GC-MS chromatogram of P. incisa methanolic extract.

No.	Compound identifier	Area	Direct parent	Molecular framework
1	(+)-Linalool	3164432	Acyclic monoterpenoids	Aliphatic acyclic compounds
2	Thujyl alcohol	6370414	Bicyclic monoterpenoids	Aliphatic homopolycyclic compounds
3	Cyclohexanone, 2-methylene-	27539848	Menthane monoterpenoids	Aliphatic homomonocyclic
	5-(1-methylethyl)-			compounds
4	Uuxrfmarqfdghe-yrnvussqsa-n	6211984	Acyclic monoterpenoids	Aliphatic acyclic compounds
5	Thymol acetate	6246479	Aromatic monoterpenoids	Aromatic homomonocyclic
				compounds
6	1,3-Cyclopentadiene, 1,2,5,5-	1936166	Branched unsaturated	Aliphatic homomonocyclic
	tetramethyl-		hydrocarbons	compounds
7	DKLRQEIICLZTQW-	3833852	Pyrazolines	Aliphatic heteromonocyclic
	UHFFFAOYSA-N			compounds
8	Pyridine, 4-methyl-, 1-oxide	3894296	Methylpyridines	Aromatic heteromonocyclic
				compounds
9	Safranal	1919330	Organic oxides	Aliphatic homomonocyclic
				compounds
10	Ethanone, 1-(2-hydroxy-4-	7888694	Alkyl-phenylketones	Aromatic homomonocyclic
	methoxyphenyl)-			compounds
11	Phenol, 4-methoxy-2,3,6-	35811264	Methoxyphenols	Aromatic homomonocyclic
	trimethyl-			compounds
12	Xanthoxylin	44718187	Alkyl-phenylketones	Aromatic homomonocyclic
				compounds
13	Bicyclo[6.1.0]non-1-ene	4810218	Polycyclic hydrocarbons	Aliphatic homopolycyclic compounds
14	5-t-Butylpyrogallol	8509713	Pyrogallols and derivatives	Aromatic homomonocyclic
				compounds
15	Mono(2-ethylhexyl) phthalate	5127844	Benzoic acid esters	Aromatic homomonocyclic
				compounds
16	4',5-Dihydroxy-7-	9018832	7-O-methylated flavonoids	Aromatic heteropolycyclic compounds
	methoxyflavanone		-	
17	3-Epimoretenol	6132472	Hopanoids	Aliphatic homopolycyclic compounds

Table 3: Phytochemical composition of *A. judaica* methanolic extract, compound peak areas, direct parents and molecular frameworks.



Fig. 3: GC-MS Chromatogram of P. undulata methanolic extract.

No	Compound identifier	Area	Direct parent	Molecular framework
1	1-	2258742	Monoterpenoids	Aliphatic homopolycyclic
	Methyltricyclo[2.2.1.0(2,6)]heptane			compounds
2	Geranylgeraniol	1426451	Acyclic diterpenoids	Aliphatic acyclic compounds
3	Bopimtnsywyzoc-uhfffaoysa-n	1920018	Eudesmane,	Aliphatic homopolycyclic
			isoeudesmane or	compounds
			cycloeudesmane	
4	Carvonhyllene oxide	6554418	Sesquiterpenoids	Alinhatic heteropolycyclic
т	Caryophynene oxide	0554410	Sesquiterpenolas	compounds
5	Farnesol (E), methyl ether	1204342	Sesquiterpenoids	Aliphatic acyclic compounds
6	alpha-Selinene	1689374	Eudesmane,	Aliphatic homopolycyclic
			isoeudesmane or	compounds
			cycloeudesmane	
7	6-Methyl-3 5-hentadiene-2-one	1216804	Enones	Aliphatic acyclic compounds
8	1-ethyl-5 5-dimethylcyclopenta-	1825261	Branched unsaturated	Aliphatic homomonocyclic
0	1,3-diene	1020201	hydrocarbons	compounds
9	2-Pentene, 2-methyl-5-nitro-	2412741	C-nitro compounds	Aliphatic acyclic compounds
10	6-Methyl-3-cyclohexen-1-	2801079	Organic oxides	Aliphatic homomonocyclic
11	carboxaldehyde	150(007	M - 4114	compounds
11	carboxylate	1300007	Wiethyl esters	compounds
12	1,6-Octadiene, 2,5-dimethyl-, (E)-	2217752	Branched unsaturated	Aliphatic acyclic compounds
			hydrocarbons	
13	2-Acetyl-5,6,7,8-	1280309	Aryl alkyl ketones	Aromatic heteropolycyclic
14	tetrahydroquinoxaline	11/6808	Albetrienes	compounds
15	Cyclocotene 3 (1 methylethenyl)	1508600	Branched unsaturated	Aliphatic homomonogyalia
15	Cyclobetene, 5-(1-methylethenyl)-	1308000	hvdrocarbons	compounds
16	(6E,11E)-6,11-Tridecadienyl	1303888	Fatty alcohol esters	Aliphatic acyclic compounds
	acetate #			
17	2,8,8-Trimethyl-4-methylene-2-	107635687	Branched unsaturated	Aliphatic homopolycyclic
18	Vinyibicycio[5.2.0]nonane # Palmitic Acid	1811981	nydrocarbons	Aliphatic acyclic compounds
10	cis-7- alpha -Bisabolene enovide	3525528	Enovides	Aliphatic beteromonocyclic
17		5525520	Epoxides	compounds
20	2,5-Dimethoxybenzoic acid	3218024	O-methoxybenzoic	Aromatic homomonocyclic
			acids and derivatives	compounds
21	Ulyktwolvhgahq-uhfffaoysa-n	5180442	Pyrroloazepines	Aromatic heteropolycyclic
22	Methyl 8 11 14-heptadecatriencate	1902493	Fatty acid methyl	Aliphatic acyclic compounds
22	Wethyr 8,11,14-heptadeeanienoae	1902493	esters	Anphane acyclic compounds
23	Benzo[g]phthalazine-1,4(2H,3H)-	57943630	Diazinanes	Aliphatic heteropolycyclic
	dione, 4a,5,6,7,8,9,10,10a-			compounds
24	octahydro-9,9-dimethyl-	7441200	Madine -1-:-	Alimbatia agralia a sur 1
24	NIEPODUAT WPJES- RMKNXTFCSA-N	/441308	aldehydes	Anphane acyclic compounds
25	3-Eicosene, (E)-	1719328	Unsaturated aliphatic	Aliphatic acyclic compounds
	- ~ /		hydrocarbons	

Table 4: Phytochemical composition of P. incisa methanolic extract, compound peak areas, direct parents and molecular frameworks.

No Compound identifier Area Direct parent Molecula	r framework
Bicyclic Aliphatic h	omopolycyclic
1 beta-Pinene 6294811 monoterpenoids com	pounds
2 Menthane Aliphatic ho	omomonocyclic
2 D-Limonene 957595 monoterpenoids com	ipounds
3 2 Isopropenyl 5 methylbey 4 engl 1401405 Acyclic Aliphatic acy	clic compounds
monoterpenoids	
4 Myrcene 813107 Acyclic Aliphatic acy	clic compounds
monoterpenoids	
Cedrane and Aliphatic he	omopolycyclic
5 Cedrene 884215 isocedrane com	pounds
Sesquiterpenoids Aliphatic h	omonolycyclic
6 Hooyrfuyendqnp-uhffaoysa-n 2468082	inounds
	pounds
7 Carvophyllene 1701118 Sesquiterpenoids Aliphatic h	omopolycyclic
com	pounds
8 alpha-Curcumene 6391414 Sesquiterpenoids, Aromatic ho	omomonocyclic
I Ioluene - Benzenoid com	pounds
9 (+)-Calarene 3645452 Aristolane Aliphatic no	omopolycyclic
10 Cadina-1(10),4-diene 7680498 Sesquiterpenoids Aliphatic ho	omopolycyclic
COM Biovelie Aliphatic h	amonolycyclic
11 Upwudjzkosqirq-uhfffaoysa-n 4763547 monoterpenoids com	inounds
Acvelic Aliphatic acv	velic compounds
12 2,6-Octadiene, 2,6-dimethyl- 11633014 monoterpenoids	••••• ••••••• • ••••••
5 10 Alimbatic h	omonolycyclic
13 Globulol 852290 cycloaromadendrane com	inounds
sesauiterpenoids	poundo
Sesquiterpenoids Aliphatic h	omopolycyclic
14 Dehydroxy-isocalamendiol 12731716 com	pounds
15 Phytol 2673801 Acyclic diterpenoids Aliphatic acy	clic compounds
Aryl thioethers, Aromatic he	omomonocyclic
Thiophenol ether com	pounds
17 n-Cymen-8-ol 2477886 Phenylpropanes Aromatic ho	omomonocyclic
com	pounds
18 Methylisoeugenol 9078334 Dimethoxybenzenes, Aromatic ho	momonocyclic
Phenol ether com	apounds
19 Spathulenol 13280356 Afoinadendrane Anphatic no	pounds
Anisoles Phenol Aromatic ho	momonocyclic
20 Isoelemicin 1679786 ethers com	pounds
Anisoles, Phenol Aromatic ho	omomonocyclic
21 Infinenyabksav-unffaoysa-n 3299054 ethers com	ipounds
22 Palmitic Acid 2211513 Long-chain fatty Aliphatic acy	clic compounds
22 Funder Ford 2211915 acids	
23 linolenic acid 907453 Lineolic acids and Aliphatic acy	clic compounds
derivatives	amanalyzzalia
DEDUDENCE SHOT APARTMENT P	Unoporycychic

 Table 5: Phytochemical composition of P. undulata methanolic extract, compound peak areas, direct parents and molecular frameworks.

Samula	Tractment	Zone of inhibition (mm) mean ± SE						
Sample	Treatment	P. aeruginosa	S. aureus	B. subtilis	E. coli			
	Methanolic extract	8.5±0.173	12±1.154	25±0.577	10±0.155			
P. undulata	Water	0.00	18.50 ± 0.088	14 ± 0.152	0.00			
	Antibiotic (AMC30)	0.00	27±0.115	20±0.1527	8±0.1154			
	Methanolic extract	9±0.1527	19.5 ± 0.252	19.5±0.265	10 ± 0.200			
P. incisa	Water	0.00	0.00	0.00	0.00			
	Antibiotic (AMC30)	0.00	30 ± 0.057	20±0.251	7 ± 0.305			
	Methanolic extract	12 ± 0.00	14 ± 0.100	12.5±0.152	10.5 ± 0.057			
A. herba-alba	Water	0.00	0.00	0.00	0.00			
	Antibiotic (AMC30)	0.00	31.5±0.251	13.5 ± 0.00	8±0.176			
1 C	Methanolic extract	$10{\pm}0.057$	14.5 ± 0.202	14 ± 0.115	10.5 ± 0.176			
A. jragrantissima	Water	0.00	0.00	0.00	0.00			
	Antibiotic (AMC30)	0.00	30 ± 0.057	17 ± 0.306	7±0.173			
1 i. J.i.	Methanolic extract	11 ± 0.00	14 ± 0.850	12 ± 057	12 ± 0.00			
A. juaaica	Water	0.00	0.00	0.00	0.00			
	Antibiotic (AMC30)	0.00	28 ± 0.360	17 ± 0.115	8.5±0.057			
	ALHM	-ALA		ARHt	>			
	ALHP	- ARHM						
	ALHtM	- ARHP						
	ALHtP	-ARHtM						

Table 6: The disc diffusion assays of the plant methanolic extracts, water and the antibiotic against the test bacteria.



Fig. 4: Stacked chart (color scheme) showing the distribution of phytochemical frameworks between plant extracts. P.u = *P. undulata*; P.i = *P. incisa*; A. j = *A. judaica*; A. h = *A. herba-alba*; A. f = *A. fragrantissima*; ALHM = aliphatic homo monocyclic; ALHP = aliphatic homo polycyclic; ALHtM = aliphatic hetero monocyclic; ALHtP = aliphatic hetero polycyclic; ALA = aliphatic acyclic; ARHM = aromatic homo monocyclic; ARHP = aromatic hetero monocyclic; ARHtP = aromatic hetero monocycli

Pak. J. Pharm. Sci., Vol.38, No.1, January-February 2025, pp.219-232

The influence of molecular framework of phytochemicals from five Asteraceae species' extracts on their antibacterial



Fig. 7: Bar chart showing the antioxidant activities (free radical scavenging ability) of plant extracts based on EDV50 values. P.u = *P. undulata*; P.i = *P. incisa*; A. j = *A. judaica*; A. h = *A. herba-alba*; A.f = *A. fragrantissima*.

ALMC -		-0.606	-0.221	0.24	-0.134	0.506	-0.106	-0.766	0.444	-0.002
ALPC -	-0.606		0.172	-0.773	-0.276	-0.64	0.552	0.673	-0.879*	-0.6
ALAC -	-0.221	0.172		-0.635	0.518	0.391	-0.32	0.009	-0.426	-0.37
ARMC -	0.24	-0.773	-0.635		-0.036	0.103	-0.357	-0.24	0.845	0.839
ARPC -	-0.134	-0.276	0.518	-0.036		0.778	0.026	-0.506	0.381	-0.237
IZ. P.a -	0.506	-0.64	0.391	0.103	0.778		-0.154	-0.889*	0.574	-0.164
IZ. S.a -	-0.106	0.552	-0.32	-0.357	0.026	-0.154		-0.072	-0.159	-0.712
IZ. B.s –	-0.766	0.673	0.009	-0.24	-0.506	-0.889*	-0.072		-0.687	0.177
IZ. E.c –	0.444	-0.879*	-0.426	0.845	0.381	0.574	-0.159	-0.687		0.482
EDV50 _	-0.002	-0.6	-0.37	0.839	-0.237	-0.164	-0.712	0.177	0.482	
	ALMC	ALPC	ALAC	ARMC	ARPC	J. 23	J. 3.8	J. 8 ^{9.}	J. 4.	4D150

ALMC = aliphatic monocyclic, ALPC = aliphatic polycyclic, ALAC = aliphatic acyclic, ARMC = aromatic monocyclic, ARPC = aromatic polycyclic. IZ.P.a = inhibition zone for *P. aeruginosa*, IZ. S.a = inhibition zone for S. aureus, IZ.B.s = inhibition zone for *B. subtilis*, IZ.E.c = inhibition zone for *E. coli*.

Fig. 8: r Pearson's correlation matrix plot (heatmap) showing the degrees and natures of association (from blue – positive, to red – negative) between concentrations of different molecular frameworks and antibacterial/ antioxidant activities of plant extracts.



Fig. 9: examples of compounds of different molecular frameworks identified in the plant extracts.

This study suggests these phytochemical frameworks and their target-specific activities be subject to further investigation to better understand their action and probable applications in pharmaceutical and medical fields. Also, studying how combining different structural frameworks can work together might help create new compounds that are better at killing bacteria and preserving cells from harm.

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